Palladium-Catalyzed Ortho-Alkoxylation of *N*-Benzoyl α -Amino Acid Derivatives at Room Temperature

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Supporting Information

ABSTRACT: An efficient palladium-catalyzed ortho-alkoxylation of *N*-benzoyl α -amino acid derivatives at room temperature has been explored. This novel transformation, using amino acids as directing groups, Pd(OAc)₂ as catalyst, alcohols as the alkoxylation reagents, and PhI(OAc)₂ as the oxidant, showed wide generality, good functional tolerance, and high monoselectivity and regioselectivity.



INTRODUCTION

Functionalized amino acid derivatives representing an important class of privileged structures have been widely found in numerous biologically active compounds and natural products (Figure 1).¹ Among them, *o*-alkoxyl-substituted *N*-benzoyl α -amino acid derivatives serve as useful intermediates in drug discovery and medicinal chemistry.² The construction of such scaffolds is mainly achieved by Ullmann reaction, while traditional methods often suffer from harsh reaction conditions, prefunctionalization of substrates, and limited generality.³ Thus, there remains the need to develop efficient and economical methodologies for elaborating amino acid derivatives under mild conditions.

In past decades, selective C–H functionalization, assisted by directing groups, has drawn considerable attention and emerged as a powerful tool to construct C–C or C–X bonds.⁴ Very recently, environmentally friendly and inexpensive amino acid moieties have been employed as novel directing groups in C–H activation for the modification of amino acid derivatives (Scheme 1A).⁵ The special structure of amino acid made itself as a feasible directing group without the necessary of being removed and the products are very useful building blocks for making bioactive molecules.

Despite the great progress that has been achieved with Pdcatalyzed direct ortho-alkoxylation of $C(sp^2)$ —H bonds,⁶ huge challenges still remain in the development of environmentally friendly and efficient transformation systems for the selective monoalkoxylation of C—H bonds under mild conditions. Given the importance of α -amino acid derivatives, we herein report the palladium-catalyzed alkoxylation of *N*-benzoyl α -amino acid derivatives at room temperature (Scheme 1B).

RESULTS AND DISCUSSION

To verify our hypothesis, we initiated investigation of the direct ortho-methoxylation of *N*-benzoyl α -amino acid **1a**. After extensive attempts, 2-(2-methoxybenzamido)-2-methylpropa-

noic acid **2a** was afforded in 29% yield with $Pd(OAc)_2$ as catalyst and DMP as oxidant at room temperature (Table 1, entry 1). As shown in Table 1, various oxidants, palladium catalysts, and solvents were screened for the best reaction conditions. The oxidant had a remarkable impact on the reaction yield, and $PhI(OAc)_2$ gave the best yield (Table 1, entry 4). The effect of different solvents on the transformation was subsequently investigated. Among them, a mixed solvent that included toluene led to a slight increase in yield (entries 5–8). Furthermore, choosing toluene as cosolvent is due to solubility. The yield was slightly decreased in the presence of oxygen, which was consistent with earlier findings (entries 10 and 11).⁷ Compared with $PdCl_2$, $Pd(OAc)_2$ proved to be the better catalyst (entry 12).

With the optimal conditions in hand, the substrate scope of *N*benzoyl α -amino acid derivatives was investigated (Table 2). Generally, various substituents both on the aromatic ring and on α -amino acid moieties were well tolerated in this direct alkoxylation reaction and afforded the corresponding monomethoxyl products in moderate to high yields (Table 2).

Cyclic amino acid derivatives worked well under standard conditions with high yields (2b-d); α -monosubstituted amino acids derivatives proceeded smoothly with moderate to high yields (2e-h), while the glycine derivative gave a slightly lower yield (2i). Notably, the chirality of the amino acids substrate was not influenced under this mild transformation condition (2h, ee > 99%), guaranteeing further applications.

The influence of different substituents at the aromatic ring was evaluated. To our delight, both electron-rich and electron-poor amino acid derivatives were well tolerated without any dimethoxyl products dectected. The protocol was found to be broadly applicable for this type of derivative bearing electron-donating or -withdrawing substituents on the phenyl ring (2j-

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Figure 1. Bioactive *o*-alkoxyl-substituted derivatives.

Scheme 1. Palladium-Catalyzed C–H Activation of Amino Acid Derivatives

A) Previous Work:



Mono-Alkoxylation

Table 1. Optimization of Reaction Conditions^a

	О N Н О Н О 1a	[Pd-cat] (10 Oxdar MeOH, sol	vent, r.t.	N OH H O Ae
entry	Pd-cat.	oxidant	solvent	yield (%)
1	$Pd(OAc)_2$	DMP	MeOH	29
2	$Pd(OAc)_2$	NaIO ₃	MeOH	NR ^b
3	$Pd(OAc)_2$	$Na_2S_2O_8$	MeOH	NR ^b
4	$Pd(OAc)_2$	$PhI(OAc)_2$	MeOH	83
5	$Pd(OAc)_2$	$PhI(OAc)_2$	DCE/MeOH(1:1)	85
6	$Pd(OAc)_2$	$PhI(OAc)_2$	CH ₃ CN/MeOH(1:1)	78
7	$Pd(OAc)_2$	$PhI(OAc)_2$	DMF/MeOH(1:1)	54
8	$Pd(OAc)_2$	$PhI(OAc)_2$	THF/MeOH(1:1)	NR ^b
9	$Pd(OAc)_2$	$PhI(OAc)_2$	PhMe/MeOH(1:1)	87
10 ^c	$Pd(OAc)_2$	$PhI(OAc)_2$	PhMe/MeOH(1:1)	76
11 ^d	$Pd(OAc)_2$	$PhI(OAc)_2$	PhMe/MeOH(1:1)	78
12	PdCl ₂	$PhI(OAc)_2$	PhMe/MeOH(1:1)	65

^{*a*}Unless otherwise specified, all reactions were carried out with 0.2 mmol of 1a, 0.02 mmol of Pd-cat., and 0.3 mmol of oxidant under argon atmosphere at room temperature for 12 h. All listed yields are isolated ones. ^{*b*}NR = no reaction. ^{*c*}The reaction were carried out under O₂ atmosphere. ^{*d*}The reaction were carried out under air.

p). Moreover, halogens, such as F, Cl, and Br, were well tolerated under the standard reaction conditions (2m-o). The broad functional group tolerance highlights the potential utility of this reaction in the late-stage modification of complex molecules as well as in the total synthesis of natural products. The cleavage of C–H bonds in meta-substituted substrates occurred predominantly at less-hindered sites to give moderate to good yields and excellent regioselectivity, irrespective of the electronic nature of the substituents (2q and 2r). The thiophene substrate also provided the desired products in moderate yield (2s). However, substrate 2-(2-methoxybenzamido)-2-methylpropanoic acid did not afford the corresponding product. The alkoxylation reaction did not proceed when the amino acid was protected with higher chain length analogues such phenylacetyl. Importantly, we also carried it out on a gram scale without any additives to afford 2a in 87% yield (Scheme 2).

Next, we investigated a variety of linear and branched alcohols as coupling partners, which demonstrated wide generality and moderate to high yields. Generally, the primary alcohols, such as ethanol, propanol, 3-trifluoro-1-propanol, and phenylethanol, could be transformed into the corresponding ethers in excellent yields (3a-d) (Table 3). The increased steric hindrance from the benzyl alcohol and branched alcohols led to a decrease in yields, while the corresponding products could be afforded in good yields at elevated temperatures (3e-i). Unfortunately, the *tert*butoxy-substituted product could not be achieved under these reaction conditions, which indicated the important influence of the steric effect.

To obtain more insight into the mechanism, some controlled experiments were performed. The addition of 2,2,6,6-tetramethylpiperidine-N-oxyl (TEMPO) as a radical quencher slightly inhibited the reaction (Scheme 3, eq 1), suggesting that the reaction does not involve a radical pathway. As shown in Scheme 3, the KIE was observed to be 4.0 (Scheme 3, eq 2), indicating that the C-H bond cleavage at the ortho-position of the Nbenzoyl α -amino acid is most likely involved with the ratelimiting step. The carboxyl group of the substrate was crucial to the reaction based on the fact that substrate 1t or 1u was not transformed into the corresponding product under the standard reaction conditions (Scheme 3, eq 3). On the basis of previous literature, ^{6c,7b,8} a plausible mechanism is proposed in Scheme 4. First, the coordination of the nitrogen atom and oxygen atom to the Pd catalyst generates a palladium intermediate (I) followed by concerted metalation-deprotonation (CMD) process to produce the palladacycle complexes II. Cyclopalladated intermediate II is then oxidized to a high-valent Pd intermediate (III) by $PhI(OAc)_2$. In the presence of alcohol solvent, the OAc ligands of III could be exchanged to form intermediate IV, which could undergo C-OR RE to give alkoxylated products.

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Table 2. Scope of N-Benzoyl α -Amino Acid Derivatives^{*a*}



^{*a*}Unless otherwise specified, all the reactions were carried out with 0.2 mmol of 1, 0.02 mmol of $Pd(OAc)_{2\nu}$ 0.3 mmol of $PhI(OAc)_{2\nu}$ and MeOH/ toluene (1:1) as solvent under argon atmosphere at room temperature for 12 h. All listed yields are isolated ones. ^{*b*}The reaction was performed at 80 °C. ^{*c*}NR = no reaction.



CONCLUSION

We have successfully developed an efficient and environmentally friendly palladium-catalyzed alkoxylation of $C(sp^2)$ –H bonds in α -amino acid derivatives at room temperature. This reaction

features broad substrate scopes, good tolerance, and high monoselectivity and regioselectivity. This mild procedure will be of importance to medicinal chemists.

EXPERIMENTAL SECTION

General Information. Unless otherwise noted, the reagents (chemicals) were purchased from commercial sources and used without further purification. Water was deionized before use. Analytical thin layer chromatography (TLC) was HSGF 254 (0.15–0.2 mm thickness). Compound spots were visualized by UV light (254 nm). Column chromatography was performed on silica gel FCP 200–300. NMR spectra were run on a 400 or 500 MHz instrument. Chemical shifts were reported in parts per million (ppm, δ) downfield from tetramethylsilane. Proton coupling patterns are described as singlet (s), doublet (d), triplet

Table 3. Scope of Various Alcohols^a



^{*a*}Unless otherwise specified, all the reactions were carried out with 0.2 mmol of 1, 0.02 mmol of $Pd(OAc)_{\mathcal{V}}$ 0.3 mmol of $PhI(OAc)_{\mathcal{V}}$ and ROH/ toluene (1:1) under argon atmosphere at room temperature for 12 h. All listed yields are isolated ones. ^{*b*}The reaction was performed at 80 °C.



(t), quartet (q), multiplet (m), and broad (br). Low- and high-resolution mass spectra (LRMS and HRMS) were measured on a spectrometer.

General Procedure for the Synthesis of Substrates 1a–s. 2-Amino-2-methylpropionic acid (2 g, 19.4 mmol) was dissolved in 1 M NaOH aqueous solution (20 mL). The mixture was cooled to 0 °C, and then benzoyl chloride (2.30 mL, 19.4 mmol) and 1 M NaOH aqueous solution (20 mL) were added dropwise simultaneously. The resulting mixture was stirred for 5 h at room temperature. Then 1 M HCl (60 mL) was added to the reaction mixture and stirred for 10 min. The resulting solid was collected by filtration and washed with water and Et₂O. The desired product 1a was obtained as a white solid (3.02 g, 72%). Compounds 1b-s were prepared in a similar manner with different yields (45–70%).

General Procedure for the Alkoxylation of Substrates. Substrate 1 (0.2 mmol), $PhI(OAc)_2$ (97 mg, 0.3 mmol), $Pd(OAc)_2$ (4.5 mg, 0.02 mmol), alcohol (1 mL), and toluene (1 mL) were added to a 25 mL tube under argon. The tube was sealed, and the mixture was stirred at room temperature for 12 h. After completion of the reaction, the solution was concentrated in a vacuum. The residue was purified using a silica gel column (DCM/MeOH/HAc = 100:5:1 as eluent) to give the corresponding pure products.

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Scheme 4. Plausible Reaction Mechanism



Determination of Intermolecular Kinetic Isotope Effect. Substrate benzoylalanine (1h) (0.2 mmol, 19 mg), (benzoyl-2,3,4,5,6- d_5)alanine (d_5 -1h) (0.2 mmol, 20 mg), PhI(OAc)₂ (97 mg, 0.3 mmol), Pd(OAc)₂ (4.5 mg, 0.02 mmol), methanol (1 mL), and toluene (1 mL) were added to a 25 mL tube under argon. The mixture was stirred at room temperature for 4 h, and the solution was concentrated by vacuum. The residue was purified with a silica gel column (DCM/MeOH/HAc = 100:5:1 as eluent) to give the corresponding pure product. The ratio of 2h/ d_4 -2h was determined to be 8.00/2.00 (KIE = 4.0) by ¹H NMR spectroscopy.

Chiral HPLC Analysis of Products. Sample A [rac-2-benzamidopropanoic acid (prepared under epimerization-free conditions from rac-2-aminopropanoic acid)], sample B [D-2-benzamidopropanoic acid (prepared under epimerization-free conditions from D-2-aminopropanoic acid), sample C synthesized from sample A (which is the product of direct C-H alkoxylation of rac-2-benzamidopropanoic acid under standard protocol)], and sample D synthesized from sample B (which is the product of direct C-H alkoxylation of D-2benzamidopropanoic acid under standard protocol)] were separated by chiral HPLC using a Chiralcel-IC column (25% *i*-PrOH and 0.5% CF₂COOH in hexanes, flow rate 0.3 mL/min, UV lamp 215 or 254 nm). In the HPLC profile of sample A, the two peaks correspond to a 1:1 mixture of D-2-benzamidopropanoic acid ($t_{\rm R}$ = 17.3 min) and L-2benzamidopropanoic acid ($t_{\rm R}$ = 20.1 min). In the HPLC profile of sample B, the peak corresponds to D-2-benzamidopropanoic acid ($t_{\rm R}$ = 17.3 min). In the HPLC profile of sample C, the peak corresponds to a 1:1 mixture of D-2-(2-methoxybenzamido)propanoic acid ($t_{\rm R}$ = 39.1 min) and L-2-(2-methoxybenzamido) propanoic acid ($t_{\rm R}$ = 48.6 min). In the HPLC profile of sample D, the peak corresponds to D-2-(2methoxybenzamido)propanoic acid, which indicates that no diastereomer of the product was observed.

Analytical Characterization Data of Products. 2-(2-Methoxybenzamido)-2-methylpropanoic Acid (1a). This compound is known.^{9a 1}H NMR (400 MHz, MeOD- d_4) δ 8.46 (s, 1H), 7.82 (d, J = 7.2 Hz, 2H), 7.55 (t, J = 7.3 Hz, 1H), 7.46 (t, J = 7.7 Hz, 2H), 1.61 (s, 6H).

1-Benzamidocyclopropane-1-carboxylic Acid (1b). This compound is known.^{9a} Compound 1b was prepared in a similar manner as described for compound 1a. ¹H NMR (400 MHz, DMSO- d_6) δ 12.36 (s, 1H), 8.95 (s, 1H), 7.86–7.82 (m, 2H), 7.52 (d, *J* = 7.3 Hz, 1H), 7.45 (t, *J* = 7.6 Hz, 2H), 1.40 (dd, *J* = 7.7, 4.4 Hz, 2H), 1.09 (dd, *J* = 7.7, 4.4 Hz, 2H).

1-Benzamidocyclopentane-1-carboxylic Acid (1c). This compound is known.^{9b} Compound 1c was prepared in a similar manner as described for compound 1a. ¹H NMR (400 MHz, DMSO- d_6) δ 12.13 (s, 1H), 8.50 (s, 1H), 7.86–7.81 (m, 2H), 7.55–7.50 (m, 1H), 7.46 (t, *J* = 7.5 Hz, 2H), 2.13 (m, 2H), 2.09–2.00 (m, 2H), 1.75–1.61 (m, 4H).

1-Benzamidocyclohexane-1-carboxylic Acid (1d). This compound is known.^{9c} Compound 1d was prepared in a similar manner as described for compound 1a. ¹H NMR (400 MHz, DMSO- d_6) δ 12.18 (s, 1H), 8.26 (s, 1H), 7.87 (d, *J* = 7.3 Hz, 2H), 7.56 (d, *J* = 6.8 Hz, 1H), 7.49 (t, *J* = 7.6 Hz, 2H), 2.16 (m, 2H), 1.83–1.70 (m, 2H), 1.57 (m, 5H), 1.32 (m, 1H).

2-Benzamido-2-phenylacetic Acid (1e). This compound is known.^{9d} Compound 1e was prepared in a similar manner as described for compound 1a. ¹H NMR (400 MHz, DMSO- d_6) δ 12.93 (s, 1H), 9.06 (d, J = 7.4 Hz, 1H), 7.93 (d, J = 7.5 Hz, 2H), 7.50 (m, 5H), 7.37 (m, 3H), 5.61 (d, J = 7.4 Hz, 1H).

Benzoylvaline (1f). This compound is known.^{9e} Compound 1f was prepared in a similar manner as described for compound 1a. ¹H NMR (400 MHz, DMSO-*d*₆) δ 12.60 (s, 1H), 8.42 (d, J = 8.2 Hz, 1H), 7.88 (d, J = 7.6 Hz, 2H), 7.54 (d, J = 7.3 Hz, 1H), 7.47 (d, J = 7.3 Hz, 2H), 4.28 (dd, J = 8.0, 6.0 Hz, 1H), 2.19 (m, 1H), 0.97 (m, 6H).

2-Benzamidobutanoic Acid (**1g**). This compound is known.^{9e} White compound **1g** was prepared in a similar manner as described for compound **1a**. ¹H NMR (400 MHz, DMSO- d_6) δ 12.56 (s, 1H), 8.55 (d, *J* = 7.6 Hz, 1H), 7.89 (d, *J* = 7.3 Hz, 2H), 7.57–7.51 (m, 1H), 7.47 (m, 2H), 4.30 (ddd, *J* = 9.2, 7.8, 5.1 Hz, 1H), 1.92–1.81 (m, 1H), 1.81–1.73 (m, 1H), 0.96 (t, *J* = 7.4 Hz, 3H).

Benzoyl-L-alanine (1*h*). This compound is known.^{9e} White compound 1h was prepared in a similar manner as described for compound 1a. ¹H NMR (400 MHz, DMSO- d_6) δ 12.53 (s, 1H), 8.65 (d, J = 7.2 Hz, 1H), 7.88 (d, J = 7.5 Hz, 2H), 7.54 (d, J = 7.3 Hz, 1H), 7.47 (m, 2H), 4.44–4.40 (m, 1H), 1.39 (d, J = 7.4 Hz, 3H).

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(Benzoyl-2,3,4,5,6- d_5)alanine (d_5 -**1h**). ¹H NMR (400 MHz, MeOD- d_4) ¹H NMR (400 MHz, MeOD- d_4) δ 4.62 (q, *J* = 7.3 Hz, 1H), 1.54 (d, *J* = 7.3 Hz, 3H).

Benzoylglycine (1i). This compound is known.^{9e} Compound 1i was prepared in a similar manner as described for compound 1a. ¹H NMR (400 MHz, MeOD- d_4) δ 7.90–7.79 (d, J = 8.4 Hz, 2H), 7.52 (d, J = 7.4 Hz, 1H), 7.49–7.37 (m, 2H), 4.09 (s, 2H).

2-Methyl-2-(4-methylbenzamido)propanoic Acid (1j). This compound is known.^{9f} Compound 1j was prepared in a similar manner as described for compound 1a. ¹H NMR (400 MHz, DMSO- d_6) δ 12.19 (s, 1H), 8.36 (s, 1H), 7.77 (d, *J* = 8.1 Hz, 2H), 7.26 (d, *J* = 8.0 Hz, 2H), 2.35 (s, 3H), 1.45 (s, 6H).

2-(4-Methoxybenzamido)-2-methylpropanoic Acid (1k). This compound is known.^{9f} Compound 1k was prepared in a similar manner as described for compound 1a. ¹H NMR (400 MHz, DMSO- d_6) δ 12.12 (s, 1H), 8.28 (s, 1H), 7.83 (d, *J* = 8.8 Hz, 2H), 6.98 (d, *J* = 8.8 Hz, 2H), 3.80 (s, 3H), 1.43 (s, 6H).

2-([1,1'-Biphenyl]-4-carboxamido)-2-methylpropanoic Acid (11). Compound 11 was prepared in a similar manner as described for compound 1a. White solid, 2.3 g, 56% yield. Mp: 156–158 °C. ¹H NMR (400 MHz, DMSO- d_6) δ 12.28 (s, 1H), 8.52 (s, 1H), 7.96 (d, *J* = 8.3 Hz, 2H), 7.77 (d, *J* = 8.2 Hz, 2H), 7.73 (d, *J* = 7.7 Hz, 2H), 7.49 (t, *J* = 7.5 Hz, 2H), 7.42 (d, *J* = 7.4 Hz, 1H), 1.47 (s, 6H). ¹³C {¹H} NMR (125 MHz, DMSO- d_6) δ 176.0, 165.9, 143.1, 139.6, 129.5, 128.6, 127.3, 126.8, 55.9, 25.5. LRMS (ESI) [M – H]⁻ *m*/*z* found: 282.0. HRMS (ESI-TOF) [M – H]⁻ *m*/*z* calcd for C₁₇H₁₆NO₃ 282.1136, found 282.1132.

2-(4-Fluorobenzamido)-2-methylpropanoic Acid (1m). Compound 1m was prepared in a similar manner as described for compound 1a. White solid, 1.6 g, 46% yield. Mp: 167–169 °C. ¹H NMR (400 MHz, DMSO- d_6) δ 12.22 (s, 1H), 8.49 (s, 1H), 7.93 (dd, *J* = 12.3, 5.3 Hz, 2H), 7.29 (dd, *J* = 12.3, 5.3 Hz, 2H), 1.45 (s, 6H). ¹³C {¹H} NMR (125 MHz, DMSO- d_6) δ 176.0, 165.2, 165.1, 163.5, 131.2, 131.1, 130.7, 130.6, 115.5, 115.4, 55.9, 25.4. LRMS (ESI) [M – H]⁻ m/z found: 224.0728, found 224.0724.

2-(4-Chlorobenzamido)-2-methylpropanoic Acid (1n). Compound 1n was prepared in a similar manner as described for compound 1a. White solid, 2.5 g, 56% yield. Mp: 185–187 °C. ¹H NMR (400 MHz, DMSO- d_6) δ 8.56 (s, 1H), 7.87 (d, *J* = 8.5 Hz, 2H, 7.53 (d, *J* = 8.5 Hz, 2H), 1.45 (s, 6H). ¹³C {¹H} NMR (125 MHz, DMSO- d_6) δ 175.96, 165.31, 136.54, 133.41, 129.90, 128.71, 56.00, 25.40. LRMS (ESI) [M – H]⁻ *m*/*z* found: 240.0. HRMS (ESI-TOF) [M – H]⁻ *m*/*z* calcd for C₁₁H₁₁ClNO₃ 240.0433, found 240.0429.

2-(4-Bromobenzamido)-2-methylpropanoic Acid (10). Compound 10 was prepared in a similar manner as described for compound 1a. White solid, 1.3 g, 55% yield. Mp: 162–165 °C. ¹H NMR (400 MHz, DMSO- d_6) δ 12.23 (s, 1H), 8.55 (s, 1H), 7.80 (dd, *J* = 8.8, 2.0 Hz, 2H), 7.68(dd, *J* = 8.8, 2.0 Hz, 2H), 1.45 (s, 6H). ¹³C {¹H} NMR (125 MHz, DMSO- d_6) δ 175.9, 165.3, 133.7, 131.6, 130.1, 125.4, 55.9, 25.4. LRMS (ESI) [M – H]⁻ *m*/*z* found: 283.9. HRMS (ESI-TOF) [M – H]⁻ *m*/*z* calcd for C₁₁H₁₁BrNO₃ 283.9928, found 283.9924.

2-Methyl-2-(4-(trifluoromethyl)benzamido)propanoic Acid (1p). Compound 1p was prepared in a similar manner as described for compound 1a. White solid, 3.1 g, 60% yield. Mp: 134–136 °C. ¹H NMR (400 MHz, DMSO- d_6) δ 12.30 (s, 1H), 8.73 (s, 1H), 8.05 (d, *J* = 8.1 Hz, 2H), 7.85 (d, *J* = 8.2 Hz, 2H), 1.47 (s, 6H). ¹³C {¹H} NMR (125 MHz, DMSO- d_6) δ 175.8, 165.2, 138.5, 131.6, 131.4, 128.8, 125.7, 125.6, 125.3, 123.5, 56.1, 25.3. LRMS (ESI) [M – H]⁻ *m*/*z* found: 274.0697, found 274.0690.

2-Methyl-2-(3-methylbenzamido)propanoic Acid (1q). Compound 1q was prepared in a similar manner as described for compound 1a. White solid, 2.4 g, 44% yield. Mp: $165-167 \,^{\circ}$ C. ¹H NMR (400 MHz, MeOD- d_4) δ 7.67 (s, 1H), 7.62 (d, $J = 6.9 \,$ Hz, 1H), 7.41–7.32 (m, 2H), 2.43 (s, 3H), 1.62 (s, 6H). ¹³C {¹H} NMR (125 MHz, DMSO- d_6) δ 176.1, 166.1, 141.5, 131.9, 129.1, 127.9, 55.8, 25.5, 21.4. LRMS (ESI) [M – H]⁻ m/z found: 220.0. HRMS (ESI-TOF) [M – H]⁻ m/z calcd for C₁₂H₁₄NO₃ 220.0979, found 220.0975.

2-Methyl-2-(3-(trifluoromethyl)benzamido)propanoic Acid (1r). Compound 1r was prepared in a similar manner as described for compound **1a**. White solid, 2.1 g, 34% yield. Mp: 145–147 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 8.71 (s, 1H), 8.11 (s, 1H), 8.05 (d, *J* = 7.8 Hz, 1H), 7.81 (d, *J* = 7.9 Hz, 1H), 7.64 (t, *J* = 7.8 Hz, 1H), 1.57 (s, 6H). ¹³C {¹H} NMR (125 MHz, DMSO- d_6) δ 175.8, 164.8, 135.4, 132.1, 130.0, 129.5, 129.3, 128.3, 125.4, 124.5, 124.4, 123.5, 56.1, 25.3. LRMS (ESI) [M – H]⁻ *m*/*z* found: 274.0. HRMS (ESI-TOF) [M – H]⁻ *m*/*z* calcd for C₁₂H₁₁F₃NO₃ 274.0697, found 274.0693.

2-Methyl-2-(thiophene-2-carboxamido)propanoic Acid (1s). Compound 1s was prepared in a similar manner as described for compound 1a. White solid, 2.3 g, 56% yield. Mp: 134–136 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 8.40 (s, 1H), 7.76 (d, *J* = 3.7 Hz, 1H), 7.64 (d, *J* = 5.0 Hz, 1H), 7.12 (dd, *J* = 4.9, 3.8 Hz, 1H), 1.58 (s, 6H). ¹³C {¹H} NMR (125 MHz, DMSO- d_6) δ 175.9, 161.1, 140.3, 131.3, 129.0, 128.3, 55.9, 25.5. LRMS (ESI) [M – H]⁻ m/z found: 212.0. HRMS (ESI-TOF) [M – H]⁻ m/z calcd for C₉H₁₀NO₃S 212.0387, found 212.0381.

2-(2-Methoxybenzamido)-2-methylpropanoic Acid (2a). This compound is known.^{9g} White solid, 41 mg, 87% yield. Mp: 142–144 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 8.99 (s, 1H), 7.91 (dd, *J* = 7.8, 1.7 Hz, 1H), 7.51–7.46 (m, 1H), 7.14 (dd, *J* = 8.3, 1.7 Hz, 1H), 7.04 (t, *J* = 8.1 Hz, 1H), 3.99 (s, 3H), 1.63 (s, 6H). ¹³C {¹H} NMR (125 MHz, MeOD- d_4) δ 178.5, 167.3, 159.7, 134.8, 132.5, 123.3, 122.5, 113.5, 58.2, 57.1, 25.6. LRMS (ESI) [M – H]⁻ m/z found: 236.0. LRMS (ESI) [M – H]⁻ m/z calcd for C₁₂H₁₄NO₄ 236.0923, found 236.0919.

1-(2-Methoxybenzamido)cyclopropane-1-carboxylic Acid (**2b**). Compound **2b** was prepared in a similar manner as described for compound **2a**. White solid, 35 mg, 75% yield. Mp: 108–110 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 7.93–7.82 (m, 3H), 7.48 (dd, *J* = 8.8, 2.2 Hz, 1H), 7.16–7.11 (m, 1H), 7.04 (t, *J* = 7.5 Hz, 1H), 3.95 (s, 3H), 1.61–1.54 (m, 2H), 1.21–1.26 (m, 2H). ¹³C {¹H} NMR (125 MHz, MeOD- d_4) δ 169.8, 159.7, 134.7, 132.6, 123.5, 122.3, 113.4, 57.0, 18.4. LRMS (ESI) [M – H]⁻ m/z found: 234.0. HRMS (ESI-TOF) [M – H]⁻ m/z calcd for C₁₂H₁₂NO₄ 234.0766, found 234.0760.

1-(2-Methoxybenzamido)cyclopentane-1-carboxylic Acid (2c). Compound 2c was prepared in a similar manner as described for compound 2a. White solid, 38 mg, 73% yield. Mp: 106–108 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 7.87 (dd, *J* = 7.7, 1.6 Hz, 1H), 7.52–7.44 (m, 1H), 7.14 (d, *J* = 8.5 Hz, 1H), 7.05 (t, *J* = 7.5 Hz, 1H), 3.98 (s, 3H), 2.25–2.34 (m, 2H), 2.14–2.07 (m, 2H), 1.80–1.89 (m, 4H). ¹³C {¹H} NMR (125 MHz, MeOD- d_4) δ 178.3, 167.9, 159.6, 134.7, 132.43, 123.4, 122.5, 113.5, 67.8, 57.1, 38.7, 26.2. LRMS (ESI) [M – H]⁻ *m*/*z* found: 262.0. HRMS (ESI-TOF) [M – H]⁻ *m*/*z* calcd for C₁₄H₁₆NO₄ 262.1079, found 262.1073.

1-(2-Methoxybenzamido)cyclohexane-1-carboxylic Acid (2d). This compound is known.^{9d} Compound 2d was prepared in a similar manner as described for compound 2a. White solid, 51 mg, 92% yield. Mp: 103–105 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 8.50 (s, 1H), 7.93–7.78 (m, 1H), 7.50 (ddd, *J* = 11.0, 8.9, 2.3 Hz, 1H), 7.17 (d, *J* = 8.6 Hz, 1H), 7.06 (t, *J* = 7.5 Hz, 1H), 4.02 (s, 3H), 2.13–2.24 (m, 2H), 1.81–1.90 (m, 2H), 1.64–1.76 (m, 4H), 1.47–1.61 (m, 2H). ¹³C {¹H} NMR (125 MHz, MeOD- d_4) δ 182.3, 171.2, 163.6, 138.8, 136.5, 126.5, 119.3, 117.5, 64.5, 61.2, 37.7, 30.8, 27.1. LRMS (ESI) [M – H]⁻ m/z found: 276.0. HRMS (ESI-TOF) [M – H]⁻ m/z calcd for C₁₅H₁₈NO₄ 276.1236, found 276.1228.

2-(2-Methoxybenzamido)-2-phenylacetic Acid (**2e**). Compound **2e** was prepared in a similar manner as described for compound **2a**. White solid, 42 mg, 73% yield. Mp: 173–175 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 7.97 (dd, *J* = 7.8, 1.6 Hz, 1H), 7.52–7.55 (m, 1H), 7.49–7.51 (m, 2H), 7.35 (t, *J* = 7.5 Hz, 2H), 7.30 (d, *J* = 7.2 Hz, 1H), 7.20 (d, *J* = 8.4 Hz, 1H), 7.06 (t, *J* = 7.5 Hz, 1H), 5.53 (s, 1H), 4.06 (s, 3H). ¹³C {¹H} NMR (125 MHz, MeOD- d_4) δ 172.4, 164.6, 157.6, 137.7, 132.7, 130.4, 127.7, 127.0, 126.3, 120.2, 120.0, 111.2, 57.4, 54.0. LRMS (ESI) [M – H]⁻ m/z found: 284.0 HRMS (ESI-TOF) [M – H]⁻ m/z calcd for C₁₆H₁₄NO₄ 284.0923, found 284.0915.

(2-Methoxybenzoyl)valine (2f). Compound 2f was prepared in a similar manner as described for compound 2a. White solid, 37 mg, 75% yield. Mp: 133–135 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 8.00–7.89 (m, 1H), 7.56–7.48 (m, 1H), 7.18 (d, *J* = 8.4 Hz, 1H), 7.08 (t, *J* = 7.6 Hz, 1H), 4.59 (d, *J* = 4.6 Hz, 1H), 4.02 (s, 3H), 2.38–2.23 (m, 1H), 1.03 (dd, *J* = 6.8, 1.4 Hz, 6H). ¹³C {¹H} NMR (125 MHz, MeOD- d_4) δ 175.6,

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167.9, 159.8, 135.0, 132.9, 122.7, 122.6, 113.6, 59.9, 57.3, 32.8, 20.1, 18.7. LRMS (ESI) $[M - H]^- m/z$ found: 250.0. HRMS (ESI-TOF) $[M - H]^- m/z$ calcd for $C_{13}H_{16}NO_4$ 250.1079 found 250.1075.

2-(2-Methoxybenzamido)butanoic Acid (**2g**). Compound **2g** was prepared in a similar manner as described for compound **2a**. White solid, 43 mg, 90% yield. Mp: 126–128 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 8.00–7.88 (m, 1H), 7.52 (ddd, *J* = 8.9, 8.0, 2.3 Hz, 1H), 7.19 (dd, *J* = 8.6, 3.2 Hz, 1H), 7.08 (t, *J* = 7.6 Hz, 1H), 4.67–4.51 (m, 1H), 4.02 (s, 3H), 2.09–1.97 (m, 1H), 1.83–1.94 (m, 1H), 1.00 (td, *J* = 7.4, 3.1 Hz, 3H). ¹³C {¹H} NMR (125 MHz, MeOD- d_4) δ 176.0, 167.8, 159.8, 135.0, 132.7, 127.6, 122.5, 113.6, 57.2, 55.9, 26.7, 10.3. LRMS (ESI) [M – H]⁻ m/z found: 236.0. HRMS (ESI-TOF) [M – H]⁻ m/z calcd for C₁₂H₁₄NO₄ 236.0923, found 236.0924.

(2-Methoxybenzoyl)-*i*-alanine (2h). Compound 2h was prepared in a similar manner as described for compound 2a. White solid, 40 mg, 89% yield. Mp: 120–122 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 8.00–7.88 (m, 1H), 7.51 (ddd, *J* = 10.2, 8.7, 2.3 Hz, 1H), 7.17 (d, *J* = 8.2 Hz, 1H), 7.07 (t, *J* = 7.6 Hz, 1H), 4.68–4.52 (m, 1H), 4.01 (d, *J* = 1.7 Hz, 3H), 1.52 (d, *J* = 7.1 Hz, 3H). ¹³C {¹H} NMR (125 MHz, MeOD- d_4) δ 167.6, 159.9, 135.0, 132.7, 122.7, 122.5, 113.5, 57.1, 19.1. LRMS (ESI) [M – H]⁻ *m*/*z* found: 222.0 HRMS (ESI-TOF) [M – H]⁻ *m*/*z* calcd for C₁₁H₁₂NO₄ 222.0766, found 222.0760.

(2-*Methoxybenzoyl)glycine* (2*i*). This compound is known.⁹¹ Compound 2*i* was prepared in a similar manner as described for compound 2*a*. White solid, 26 mg, 62% yield. Mp: 117–119 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 8.00 (dd, J = 7.8, 1.8 Hz, 1H), 7.51 (ddd, J = 8.4, 7.3, 1.9 Hz, 1H), 7.14 (d, J = 8.4 Hz, 1H), 7.06 (d, J = 8.2 Hz, 1H), 4.14 (s, 2H), 3.99 (s, 3H). ¹³C {¹H} NMR (101 MHz, MeOD- d_4) δ 173.6, 168.3, 159.9, 135.0, 132.8, 122.4, 113.4, 57.0, 43.1. LRMS (ESI) [M – H]⁻ m/z found: 208.0. HRMS (ESI-TOF) [M – H]⁻ m/z calcd for C₁₀H₁₀NO₄ 208.0610, found 208.0607.

2-(2-Methoxy-4-methylbenzamido)-2-methylpropanoic Acid (2j). Compound 2j was prepared in a similar manner as described for compound 2a. White solid, 34 mg, 67% yield. Mp: 98–100 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 8.90 (s, 1H), 7.84 (d, J = 7.9 Hz, 1H), 6.98 (s, 1H), 6.89 (d, J = 8.0 Hz, 1H), 4.00 (s, 3H), 2.40 (s, 3H), 1.65 (s, 6H). ¹³C {¹H} NMR (150 MHz, MeOD- d_4) δ 165.3, 157.8, 144.1, 130.7, 121.4, 118.5, 114.4, 112.2, 61.2, 55.1, 23.7, 20.3. LRMS (ESI) [M – H]⁻ m/z found: 250.0. HRMS (ESI-TOF) [M – H]⁻ m/z calcd for C₁₃H₁₆NO₄ 250.1079, found 250.1029.

2-(2,4-Dimethoxybenzamido)-2-methylpropanoic Acid (2k). Compound 2k was prepared in a similar manner as described for compound 2a. White solid, 38 mg, 72% yield. Mp: 112–114 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 7.91 (d, *J* = 8.9, 1H), 6.63 (m, 2H), 3.99 (s, 3H), 3.85 (s, 3H), 1.64 (s, 6H). ¹³C {¹H} NMR (100 MHz, MeOD- d_4) δ 178.6, 166.9, 165.9, 161.2, 134.3, 115.6, 107.3, 99.9, 57.12, 56.5, 25.7. LRMS (ESI) [M – H]⁻ *m*/*z* found: 266.0. HRMS (ESI-TOF) [M – H]⁻ *m*/*z* calcd for C₁₃H₁₆NO₅ 266.1028, found 266.1024.

2-(3-Methoxy-[1,1'-biphenyl]-4-carboxamido)-2-methylpropanoic Acid (2l). Compound 2l was prepared in a similar manner as described for compound 2a. White solid, 34 mg, 55% yield. Mp: 156– 158 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 9.01 (s, 1H), 8.01 (d, *J* = 8.1 Hz, 1H), 7.68 (dd, *J* = 8.3, 1.3 Hz, 2H), 7.48–7.44 (m, 2H), 7.39 (d, *J* = 7.4 Hz, 1H), 7.34 (d, *J* = 1.5 Hz, 1H), 7.32 (dd, *J* = 8.1, 1.6 Hz, 1H), 4.08 (s, 3H), 1.66 (s, 6H). ¹³C {¹H} NMR (100 MHz, MeOD- d_4) δ 166.9, 160.07, 148.0, 141.7, 133.2, 130.5, 129.7, 128.7, 122.1, 122.0, 121.0, 111.9, 58.5, 57.2, 25.6. LRMS (ESI) [M – H]⁻ m/z found: 312.0. HRMS (ESI-TOF) [M – H]⁻ m/z calcd for C₁₈H₁₈NO₄ 312.1236, found 312.1234.

2-(4-Fluoro-2-methoxybenzamido)-2-methylpropanoic Acid (2*m*). Compound 2*m* was prepared in a similar manner as described for compound 2*a*. White solid, 33 mg, 64% yield. Mp: 86–88 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 8.81 (s, 1H), 7.97 (dd, *J* = 8.6, 7.1 Hz, 1H), 6.97 (dd, *J* = 11.0, 2.3 Hz, 1H), 6.82 (td, *J* = 8.3, 2.3 Hz, 1H), 4.02 (s, 3H), 1.65 (s, 6H). ¹³C {¹H} NMR (125 MHz, MeOD- d_4) δ 174.7, 163.5, 157.3, 130.6, 115.9, 105.1, 97.5, 54.3, 53.6, 21.6. LRMS (ESI) [M – H]⁻ *m*/*z* calcd for C₁₂H₁₃NO₄F 254.0829, found 254.0829.

2-(4-Chloro-2-methoxybenzamido)-2-methylpropanoic Acid (2n). Compound 2n was prepared in a similar manner as described for compound **2a.** White solid, 30 mg, 55% yield. Mp: 112–114 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 8.91 (s, 1H), 7.90 (d, *J* = 8.4 Hz, 1H), 7.21 (d, *J* = 1.8 Hz, 1H), 7.09 (dd, *J* = 8.4, 1.8 Hz, 1H), 4.03 (s, 3H), 1.65 (s, 6H). ¹³C {¹H} NMR (125 MHz, MeOD- d_4) δ 165.0, 159.0, 138.9, 132.6, 121.4, 121.3, 112.9, 56.4, 24.3. LRMS (ESI) [M – H]⁻ *m*/*z* found: 270.0. HRMS (ESI-TOF) [M – H]⁻ *m*/*z* calcd for C₁₂H₁₃NO₄Cl 270.0533, found 270.0531.

2-(4-Bromo-2-methoxybenzamido)-2-methylpropanoic Acid (20). Compound 20 was prepared in a similar manner as described for compound 2a. White solid, 39 mg, 62% yield. Mp: 116–118 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 8.90 (s, 1H), 7.81 (d, *J* = 8.4 Hz, 1H), 7.33 (s, 1H), 7.23 (dd, *J* = 8.3, 1.4 Hz, 1H), 4.02 (s, 3H), 1.65 (s, 6H). ¹³C {¹H} NMR (100 MHz, MeOD- d_4) δ 166.3, 160.0, 133.9, 128.3, 125.7, 122.8, 117.0, 58.6, 57.6, 25.5. LRMS (ESI) [M – H]⁻ *m*/*z* found: 314.0. HRMS (ESI-TOF) [M – H]⁻ *m*/*z* calcd for C₁₂H₁₃BrNO₄ 314.0028, found 314.0019, 316.003.

2-(2-Methoxy-4-(trifluoromethyl)benzamido)-2-methylpropanoic Acid (**2p**). Compound **2p** was prepared in a similar manner as described for compound **2a**. White solid, 35 mg, 57% yield. Mp: 160–162 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 8.99 (s, 1H), 8.03 (d, *J* = 8.0 Hz, 1H), 7.40 (s, 1H), 7.37 (d, *J* = 8.3 Hz, 1H), 4.07 (s, 3H), 1.66 (s, 6H). ¹³C {¹H} NMR (125 MHz, MeOD- d_4) δ 177.7, 164.9, 158.4, 134.52, 132.0, 126.6, 124.2, 117.7, 109.23,57.5, 56.4, 24.3. LRMS (ESI) $[M - H]^- m/z$ found: 304.0. HRMS (ESI-TOF) $[M - H]^- m/z$ calcd for C₁₃H₁₃NO₄F₃ 304.0797, found 304.0793.

2-(2-Methoxy-5-methylbenzamido)-2-methylpropanoic Acid (2q). Compound 2q was prepared in a similar manner as described for compound 2a. White solid, 44 mg, 88% yield. Mp: 142–144 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 8.93 (s, 1H), 7.75 (d, J = 2.0 Hz, 1H), 7.31 (dd, J = 8.4, 2.2 Hz, 1H), 7.04 (d, J = 8.4 Hz, 1H), 3.98 (s, 3H), 1.65 (s, 6H). ¹³C {¹H} NMR (125 MHz, MeOD- d_4) δ 178.8, 167.31, 157.6, 135.1, 132.7, 131.9, 122.8, 113.5, 58.3, 57.1, 25.6, 20.9. LRMS (ESI) [M – H]⁻ m/z found: 250.0. HRMS (ESI-TOF) [M – H]⁻ m/z calcd for C₁₃H₁₆NO₄ 250.1079, found 250.1074.

2-(2-Methoxy-5-(trifluoromethyl)benzamido)-2-methylpropanoic Acid (2r). Compound 2r was prepared in a similar manner as described for compound 2a. White solid, 37 mg, 60% yield. Mp: 177–179 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 8.19 (d, J = 2.1 Hz, 1H), 7.81 (dd, J = 8.7, 2.1 Hz, 1H), 7.34 (d, J = 8.7 Hz, 1H), 4.09 (s, 3H), 1.66 (s, 6H). ¹³C {¹H} NMR (125 MHz, MeOD- d_4) δ 178.6, 165.8, 162.0, 131.4, 129.4, 126.1, 124.4, 124.4, 114.2, 58.5, 57.6, 25.5. LRMS (ESI) [M – H]⁻ m/z found: 304.0. HRMS (ESI-TOF) [M – H]⁻ m/z calcd for C₁₃H₁₃NO₄F₃ 304.0797, found 304.0790.

2-(3-Methoxythiophene-2-carboxamido)-2-methylpropanoic Acid (2s). Compound 2s was prepared in a similar manner as described for compound 2a. White solid, 24 mg, 49% yield. Mp: 137–139 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 8.13 (s, 2H), 7.61 (d, *J* = 5.5 Hz, 1H), 7.08 (d, *J* = 5.5 Hz, 1H), 4.08 (s, 3H), 1.65 (s, 6H). ¹³C {¹H} NMR (125 MHz, MeOD- d_4) δ 178.5, 163.6, 159.4, 131.1, 117.6, 117.2, 60.3, 58.1, 25.7. LRMS (ESI) [M – H]⁻ *m*/*z* found: 242.0 HRMS (ESI-TOF) [M – H]⁻ *m*/*z* calcd for C₁₀H₁₂NO₄S 242.0487, found 242.0482.

2-(2-Ethoxybenzamido)-2-methylpropanoic Acid (**3a**). Compound **3a** was prepared in a similar manner as described for compound **2a**. White solid, 49 mg, 95% yield. Mp: 137–139 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 9.02 (s, 1H), 7.96 (d, *J* = 7.7 Hz, 1H), 7.50 (t, *J* = 7.5 Hz 2H), 7.14 (d, *J* = 8.4 Hz, 1H), 7.06 (t, *J* = 7.6 Hz, 1H), 4.26 (q, *J* = 6.9 Hz, 2H), 1.66 (s, 6H), 1.56 (t, *J* = 7.0 Hz, 3H). ¹³C {¹H} NMR (125 MHz, MeOD- d_4) δ 167.2, 159.1, 134.7, 132.5, 123.2, 122.4, 114.40, 66.6, 25.7, 15.6. LRMS (ESI) [M – H]⁻ m/z found: 250.0 HRMS (ESI-TOF) [M – H]⁻ m/z calcd for C₁₃H₁₆NO₄ 250.1079, found 250.1078.

2-Methyl-2-(2-propoxybenzamido)propanoic Acid (**3b**). Compound **3b** was prepared in a similar manner as described for compound **2a**. White solid, 50 mg, 94% yield. Mp: 167–169 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 8.86 (s, 1H), 7.92 (dd, *J* = 7.8, 1.7 Hz, 1H), 7.55–7.33 (m, 1H), 7.12 (d, *J* = 8.3 Hz, 1H), 7.03 (t, *J* = 7.6 Hz, 1H), 4.13 (t, *J* = 6.4 Hz, 2H), 1.99–1.88 (m, 2H), 1.63 (s, 6H), 1.10 (t, *J* = 7.4 Hz, 3H). ¹³C {¹H} NMR (125 MHz, MeOD- d_4) δ 178.5, 167.3, 159.1, 134.8, 132.5, 123.2, 122.3, 114.3, 72.5, 58.2, 25.9, 24.0, 11.6. LRMS (ESI) [M – H]⁻ *m*/*z* found: 264.0. HRMS (ESI-TOF) [M – H]⁻ *m*/*z* calcd for C₁₄H₁₈NO₄ 264.1236, found 264.1240.

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2-Methyl-2-(2-(3,3,3-trifluoropropoxy)benzamido)propanoic Acid (3c). Compound 3c was prepared in a similar manner as described for compound 2a. White solid, 53 mg, 83% yield. Mp: 175–177 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 8.74 (s, 1H), 7.91 (dd, *J* = 7.8, 1.4 Hz, 1H), 7.56–7.43 (m, 1H), 7.15 (d, *J* = 8.3 Hz, 1H), 7.09 (t, *J* = 7.6 Hz, 1H), 4.41 (t, *J* = 5.9 Hz, 2H), 2.96–2.80 (m, 1H), 1.64 (s, 6H). ¹³C {¹H} NMR (125 MHz, MeOD- d_4) δ 178.8, 167.0, 158.1, 134.7, 132.7, 123.9, 123.0, 114.0, 63.9, 58.5, 34.9, 25.4. LRMS (ESI) [M – H][–] *m*/*z* found: 318.0. HRMS (ESI-TOF) [M – H]– *m*/*z* calcd for C₁₄H₁₅NO₄F₃ 318.0953, found 318.0945.

2-Methyl-2-(2-phenethoxybenzamido)propanoic Acid (**3d**). Compound **3d** was prepared in a similar manner as described for compound **2a**. White solid, 50 mg, 76% yield. Mp: 183–185 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 8.73 (s, 1H), 7.93 (d, J = 7.7 Hz, 1H), 7.50 (t, J = 7.0 Hz, 1H), 7.38–7.27 (m, 4H), 7.26–7.21 (m, 1H), 7.19 (d, J = 8.5 Hz, 1H), 7.07 (t, J = 7.6 Hz, 1H), 4.48 (t, J = 6.7 Hz, 2H), 3.27 (t, J = 6.7 Hz, 2H), 1.51 (s, 6H). ¹³C {¹H} NMR (125 MHz, MeOD- d_4) δ 167.1, 158.7, 139.7, 134.7, 132.7, 130.3, 130.3, 130.1, 128.1, 123.4, 122.5, 114.4, 70.9, 36.5, 25.6. LRMS (ESI) [M – H]⁻ m/z found: 326.0. HRMS (ESI-TOF) [M – H]⁻ m/z calcd for C₁₉H20NO₄ 326.1392, found 326.1393.

2-(2-(Benzyloxy)benzamido)-2-methylpropanoic Acid (**3e**). Compound **3e** was prepared in a similar manner as described for compound **2a**. White solid, 29 mg, 46% yield. Mp: 178–180 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 8.67 (s, 1H), 7.95 (d, J = 7.7 Hz, 1H), 7.58 (d, J = 7.1 Hz, 2H), 7.47–7.54 (m, 1H), 7.40–7.49 (m, 2H), 7.38–7.44 (m, 1H), 7.28 (d, J = 8.3 Hz, 1H), 7.10 (t, J = 7.6 Hz, 1H), 5.27 (s, 2H), 1.38 (s, 6H). ¹³C {¹H} NMR (125 MHz, MeOD- d_4) δ 167.2, 158.9, 137.9, 134.7, 132.6, 130.3, 130.1, 130.1, 123.6, 122.7, 114.9, 73.0, 25.8. LRMS (ESI) [M – H]⁻ m/z found: 312.0. HRMS (ESI-TOF) [M – H]⁻ m/z calcd for C₁₈H₁₈NO₄ 312.1236, found 312.1230.

2-(2-Isobutoxybenzamido)-2-methylpropanoic Acid (**3f**). Compound **3f** was prepared in a similar manner as described for compound **2a**. White solid, 40 mg, 72% yield. Mp: 143–145 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 8.77 (s, 1H), 7.92 (dd, *J* = 7.8, 1.7 Hz, 1H), 7.54–7.44 (m, 1H), 7.14 (d, *J* = 8.4 Hz, 1H), 7.06 (t, *J* = 7.5 Hz, 1H), 3.97 (d, *J* = 6.5 Hz, 2H), 2.18–2.30 (m, 1H), 1.65 (s, 6H), 1.11 (d, *J* = 6.1 Hz, 6H). ¹³C {¹H} NMR (125 MHz, MeOD- d_4) δ 178.4, 167.5, 159.1, 134.7, 132.5, 123.3, 122.3, 114.2, 77.2, 58.1, 29.9, 25.9, 20.2. LRMS (ESI) [M – H]⁻ *m*/*z* found: 278.0. HRMS (ESI-TOF) [M – H]⁻ *m*/*z* calcd for C₁₅H₂₀NO₄ 278.1392, found 278.1385.

2-(2-Isopropoxybenzamido)-2-methylpropanoic Acid (**3g**). Compound **3g** was prepared in a similar manner as described for compound **2a**. White solid, 26 mg, 49% yield. Mp: 137–139 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 9.00 (s, 1H), 7.94 (dd, J = 7.8, 1.7 Hz, 1H), 7.51–7.38 (m, 1H), 7.15 (d, J = 8.4 Hz, 1H), 7.03 (t, J = 7.6 Hz, 1H), 4.82–4.91 (m, 1H), 1.64 (s, 6H), 1.44 (d, J = 6.1 Hz, 6H). ¹³C {¹H} NMR (125 MHz, MeOD- d_4) δ 176.5, 165.5, 165.4, 156.1, 132.7, 130.8, 122.3, 120.6, 114.3, 72.1, 23.8, 20.9. LRMS (ESI) [M – H]⁻ m/z found: 264.0. HRMS (ESI-TOF) [M – H]⁻ m/z calcd for C₁₄H₁₈NO₄ 264.1236, found 264.1229.

2-(2-(Cyclopropylmethoxy)benzamido)-2-methylpropanoic Acid (**3h**). Compound **3h** was prepared in a similar manner as described for compound **2a**. White solid, 19 mg, 35% yield. Mp: 146–148 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 9.05 (s, 2H), 7.97 (dd, *J* = 7.8, 1.7 Hz, 1H), 7.51–7.43 (m, 1H), 7.13–6.99 (m, 2H), 4.03 (d, *J* = 7.2 Hz, 2H), 1.65 (s, 6H), 1.37–1.47 (m, 1H), 0.73–0.66 (m, 2H), 0.43–0.56 (m, 2H). ¹³C {¹H} NMR (125 MHz, MeOD- d_4) δ 178.5, 167.2, 159.2, 134.9, 132.6, 122.9, 122.4, 114.5, 75.8, 58.0, 26.0, 11.5, 4.3. LRMS (ESI) [M – H]⁻ *m*/*z* found: 276.0. HRMS (ESI-TOF) [M – H]⁻ *m*/*z* calcd for C₁₅H₁₈NO₄ 276.1236, found 276.1231.

2-(2-(Cyclohexyloxy)benzamido)-2-methylpropanoic Acid (**3***i*). Compound **3***i* was prepared in a similar manner as described for compound **2a**. White solid, 31 mg, 50% yield. Mp: 150–152 °C. ¹H NMR (400 MHz, MeOD- d_4) δ 8.96 (s, 1H), 7.94–7.88 (dd, *J* = 8.4, 1.2, 1H), 7.45 (t, *J* = 7.8 Hz, 1H), 7.16 (d, *J* = 8.3 Hz, 1H), 7.02 (t, *J* = 7.5 Hz, 1H), 4.59–4.49 (m, 1H), 2.04–2.14(m, 2H), 1.79–1.89 (m, 2H), 1.62–1.68 (m, 8H), 1.50–1.43 (m, 2H), 1.42–1.32 (m, 2H). ¹³C {¹H} NMR (100 MHz, MeOD- d_4) δ 167.5, 157.7, 134.4, 132.7, 124.6, 122.3, 116.0, 79.1, 33.4, 27.0, 25.8, 25.4. LRMS (ESI) [M – H][–] *m*/*z* found: 304.0. HRMS (ESI-TOF) $[M - H]^- m/z$ calcd for $C_{17}H_{22}NO_4$ 304.1549, found 304.1545.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.6b02257.

Copies of ¹H and ¹³C {¹H} NMR spectra and HPLC experiments data (PDF)

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Notes

The authors declare no competing financial interest.

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REFERENCES

 (1) For selected papers, see: (a) Majmudar, J. D.; Hodges-Loaiza, H. B.; Hahne, K.; Donelson, J. L.; Song, J.; Shrestha, L.; Harrison, M. L.; Hrycyna, C. A.; Gibbs, R. A. *Bioorg. Med. Chem.* **2012**, *20*, 283–295.
 (b) Roughley, S. D.; Jordan, A. M. *J. Med. Chem.* **2011**, *54*, 3451–3479.
 (c) Takeuchi, Y.; Ozaki, S.; Satoh, M.; Mimura, K.; Hara, S.; Abe, H.; Nishioka, H.; Harayama, T. *Chem. Pharm. Bull.* **2010**, *58*, 1552–1553.
 (d) Wang, S.; Beck, R.; Blench, T.; Burd, A.; Buxton, S.; Malic, M.; Ayele, T.; Shaikh, S.; Chahwala, S.; Chander, C.; Holland, R.; Merette, S.; Zhao, L.; Blackney, M.; Watts, A. *J. Med. Chem.* **2010**, *53*, 1465–1472.

(2) (a) Imramovsky, A.; Jorda, R.; Pauk, K.; Reznickova, E.; Dusek, J.; Hanusek, J.; Krystof, V. *Eur. J. Med. Chem.* **2013**, *68*, 253–259. (b) Yang, B.; Lamb, M. L.; Zhang, T.; Hennessy, E. J.; Grewal, G.; Sha, L.; Zambrowski, M.; Block, M. H.; Dowling, J. E.; Su, N.; Wu, J.; Deegan, T.; Mikule, K.; Wang, W.; Kaspera, R.; Chuaqui, C.; Chen, H. *J. Med. Chem.* **2014**, *57*, 9958–9970.

(3) (a) Maier, T. C.; Fu, G. C. J. Am. Chem. Soc. 2006, 128, 4594–4595.
(b) Shintou, T.; Mukaiyama, T. J. Am. Chem. Soc. 2004, 126, 7359–7367. (c) Vo, C. V.; Mitchell, T. A.; Bode, J. W. J. Am. Chem. Soc. 2011, 133, 14082–14089.

(4) (a) Ackermann, L. Chem. Rev. 2011, 111, 1315–1345. (b) Engle, K. M.; Mei, T. S.; Wasa, M.; Yu, J. Q. Acc. Chem. Res. 2012, 45, 788–802. (c) Peron, F.; Fossey, C.; Sopkova-de Oliveira Santos, J.; Cailly, T.; Fabis, F. Chem. - Eur. J. 2014, 20, 7507–7513. (d) Bag, S.; Patra, T.; Modak, A.; Deb, A.; Maity, S.; Dutta, U.; Dey, A.; Kancherla, R.; Maji, A.; Hazra, A.; Bera, M.; Maiti, D. J. Am. Chem. Soc. 2015, 137, 11888–11891. (e) Bera, M.; Maji, A.; Sahoo, S. K.; Maiti, D. Angew. Chem., Int. Ed. 2015, 54, 8515–8519. (f) Kolle, S.; Batra, S. Org. Biomol. Chem. 2015, 13, 10376–10385. (g) Maji, A.; Bhaskararao, B.; Singha, S.; Sunoj, R. B.; Maiti, D. Chem. Sci. 2016, 7, 3147–3153. (h) Bera, M.; Sahoo, S. K.; Maiti, D. ACS Catal. 2016, 6, 3575–3579.

(5) (a) Castro, L. C. M.; Chatani, N. *Chem. - Eur. J.* **2014**, *20*, 4548–4553. (b) Gong, W.; Zhang, G.; Liu, T.; Giri, R.; Yu, J. Q. J. Am. Chem. Soc. **2014**, *136*, 16940–16946. (c) Kim, J.; Sim, M.; Kim, N.; Hong, S. Chem. Sci. **2015**, *6*, 3611–3616. (d) Toba, T.; Hu, Y.; Tran, A. T.; Yu, J. Q. Org. Lett. **2015**, *17*, 5966–5969.

(6) (a) Dick, A. R.; Hull, K. L.; Sanford, M. S. J. Am. Chem. Soc. 2004, 126, 2300–2301. (b) Gao, T.; Sun, P. J. Org. Chem. 2014, 79, 9888–9893. (c) Jiang, T. S.; Wang, G. W. J. Org. Chem. 2012, 77, 9504–9509. (d) Shi, S.; Kuang, C. J. Org. Chem. 2014, 79, 6105–6112. (e) Siciliano,

The Journal of Organic Chemistry

C.; Barattucci, A.; Bonaccorsi, P.; Di Gioia, M. L.; Leggio, A.; Minuti, L.; Romio, E.; Temperini, A. J. Org. Chem. **2014**, 79, 5320–5326. (f) Yin, Z.; Jiang, X.; Sun, P. J. Org. Chem. **2013**, 78, 10002–10007. (g) Zhang, C.; Sun, P. J. Org. Chem. **2014**, 79, 8457–8461. (h) Zhang, L. B.; Hao, X. Q.; Zhang, S. K.; Liu, K.; Ren, B.; Gong, J. F.; Niu, J. L.; Song, M. P. J. Org. Chem. **2014**, 79, 10399–10409.

(7) (a) Chen, F.-J.; Zhao, S.; Hu, F.; Chen, K.; Zhang, Q.; Zhang, S.-Q.; Shi, B.-F. *Chem. Sci.* **2013**, *4*, 4187–4192. (b) Zhang, S. Y.; He, G.; Zhao, Y.; Wright, K.; Nack, W. A.; Chen, G. J. Am. Chem. Soc. **2012**, *134*, 7313–7316.

(8) (a) Li, H.; Li, P.; Wang, L. Org. Lett. **2013**, 15, 620–623. (b) Lian, B.; Zhang, L.; Chass, G. A.; Fang, D. C. J. Org. Chem. **2013**, 78, 8376–8385.

(9) (a) Lucchesi, C.; Arbore, A.; Pascual, S.; Fontaine, L.; Maignan, C.; Dujardin, G. Carbohydr. Res. **2010**, 345, 844–849. (b) Satyanarayana, B.; Sumalatha, Y.; Sridhar, C.; Venkatraman, S.; Reddy, P. P. *Heterocycl. Commun.* **2006**, *12*, 323–328. (c) Saavedra, C.; Hernandez, R.; Boto, A.; Alvarez, E. J. Org. Chem. **2009**, *74*, 4655–4665. (d) Sofia; Chakravarty; Katzenellenbogen. J. Org. Chem. **1983**, *48*, 3318–3325. (e) Saavedra, C.; Hernandez, R.; Boto, A.; Alvarez, E. J. Org. Chem. **2009**, *74*, 4655–4665. (f) Reinaud; Capdevielle; Maumy. Synthesis **1990**, *7*, 612–614. (g) Buijs, W.; Comba, P.; Corneli, D.; Pritzkow, H. J. Organomet. Chem. **2002**, *641*, 71–80. (h) Schaefer, G.; Bode, J. W. Org. Lett. **2014**, *16*, 1526–1529. (i) Su, W.; Chen, J.; Jin, C. Heterocycles **2011**, *83*, 153–161.